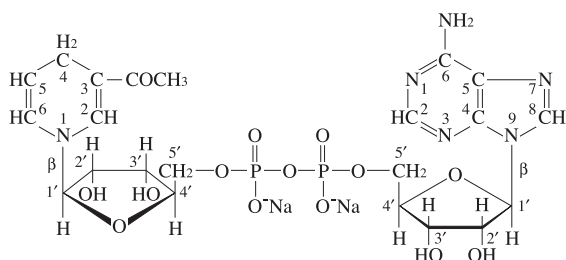


# APADH

3-Acetylpyridine-adenine dinucleotide, reduced form (disodium salt)

*prepared enzymatically*

## Structure



**Formula** :  $C_{22}H_{28}N_6O_{14}P_2 \cdot Na_2$

**Formula weight** : 708.4

## Specification

### Purity

Determined by Enzymatic Method (ADH)

### Water Content

### Na

### UV Spectral Analysis

Ratio at pH 7.5

$$A_{250}/A_{260}$$

$$A_{280}/A_{260}$$

### Specifications

$\geq 92\%$

$< 8\%$

$6.0 \pm 2\%$

$0.82 \pm 0.04$

$0.23 \pm 0.03$

## Assay Procedure

### I. Spectrophotometric Method

Wavelength ; 363 nm, Light path length ; 1 cm

Pipette the following reagents into a cuvette

|                      | a      | b      | c      | d      |
|----------------------|--------|--------|--------|--------|
| Acetaldehyde buffer* | 5.0 mL | 5.0 mL | 5.0 mL | 5.0 mL |
| ADH (1 IU/mL)        | 0.2 mL | —      | 0.2 mL | —      |
| APADH (0.4 mg/mL)    | 0.5 mL | 0.5 mL | —      | —      |
| Distilled water      | 0.3 mL | 0.5 mL | 0.8 mL | 1.0 mL |

\* 83.3 mmol/L Tris-HCl, pH 7.5 containing 34 mmol/L acetoaldehyde

### II. Calculation

$$\frac{\Delta A \cdot V \cdot MW \times 100}{9.1 \times 10^3 \cdot d \cdot v \cdot s} \times \frac{100}{(100 - S - W)} = \text{Purity of APADH}$$

$$\Delta A = A_b - (A_a + A_c)$$

V = Total volume of reaction mixture (6.0 mL)

MW = 664.4, as of anhydrate

$9.1 \times 10^3$  = Molar extinction coefficient of APADH at 363 nm ( $L \cdot mol^{-1} \cdot cm^{-1}$ )

d = Light path length (1 cm)

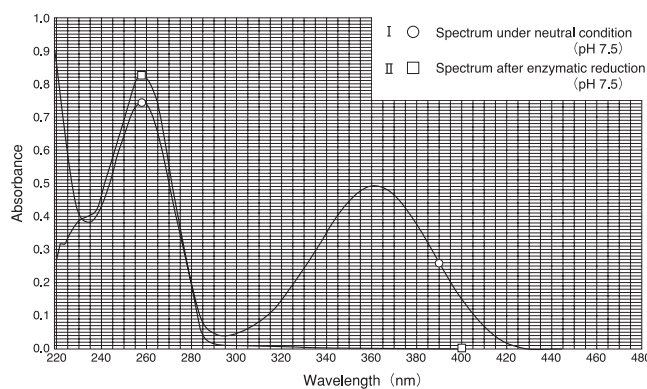
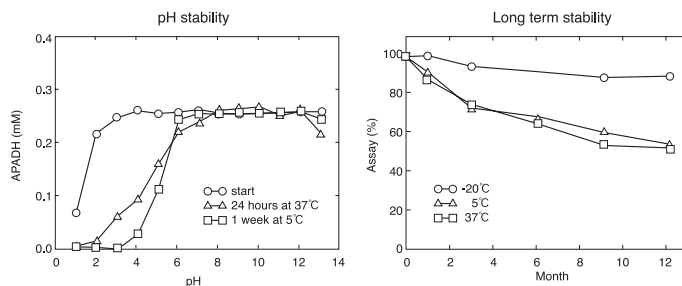
v = Sample volume (0.5 mL)

s = Sample concentration (0.4 mg/mL)

S = Na (%)

W = Water Content (%)

## Reference Data



## Storage

Keep tightly stoppered in the dark below 5°C.

Moisture will accelerate the purity reduction.

For prolonged storage keep below -20°C.

## OYC No./Package

| OYC No.  | Package |
|----------|---------|
| 44049000 | 100 mg  |
| 44048900 | Bulk    |

(Research reagent use only, not for medical use.)



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